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Oxidative Organosolv Fractionation of Lignocellulosic Biomass Assisted by Solid Catalysis and Biochemical Sugars Conversion

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1 INTRODUCTION

► Biorefineries aim to convert renewable lignocellulosic biomass to sustainable fuels and chemical products.

► A crucial step in this process is the **pretreatment of the biomass**, which separates it into cellulose, hemicellulose, and lignin streams.

► CERTH has developed an **innovative organosolv pretreatment method known as OxiOrganosolv (OOS)**.

► OOS effectively fractionates lignocellulosic biomass under oxidising conditions, eliminating the need for soluble acid catalysts typically required in conventional organosolv processes. The optimal conditions for this method were 175 °C, 2 h, a 50:50 water-to-organic solvent ratio, and a 16 bar, 100% oxygen atmosphere [1].

Objective

► Reduce the intensity of the OOS conditions to **make the process more energy-efficient and cost-effective**.

Approach

► We utilised **solid acid catalysts (zeolites)** to lower the reaction temperature to 150 °C.

► We screened various zeolite catalysts.

► We studied the catalyst-to-feed (C/F) ratio effect with the optimum zeolite.

► We **recovered** and characterised the catalyst to investigate changes in its properties

2 EXPERIMENTAL

Feedstock: Hard wheat straw (WS) collected from fields in Northern Greece

Feedstock	Extractives, wt. %	Cellulose, wt. %	Hemicellulose, wt. %	Lignin, wt. %
Wheat straw	15.5	39.9	22.9	16.6

Catalysts: Four different zeolitic types (Mordenite, Y, Beta and ZSM-5) with varying Al content (SAR: 5-300)

Catalyst*	Surface area, m ² /g	Micropore surface area, m ² /g	Brønsted acid sites, μmol/g	Lewis acid sites, μmol/g	Total acid sites, μmol/g
H-Y(5)	880	764	385	146	531
H-Y(12)	870	713	263	184	447
H-Y(30)	885	641	184	83	267
H-Y(60)	908	610	118	32	150
H-Beta(25)	609	423	229	177	406
H-Beta(75)	679	443	133	81	214
H-Beta(300)	651	488	68	5	73
H-MOR(20)	540	478	384	142	526
H-ZSM5(30)	410	260	214	87	301

*The silica-to-alumina (SAR) ratio is indicated in parentheses

Experimental setup: OOS runs were carried out in a 975 mL autoclave reactor (Fig. 1).

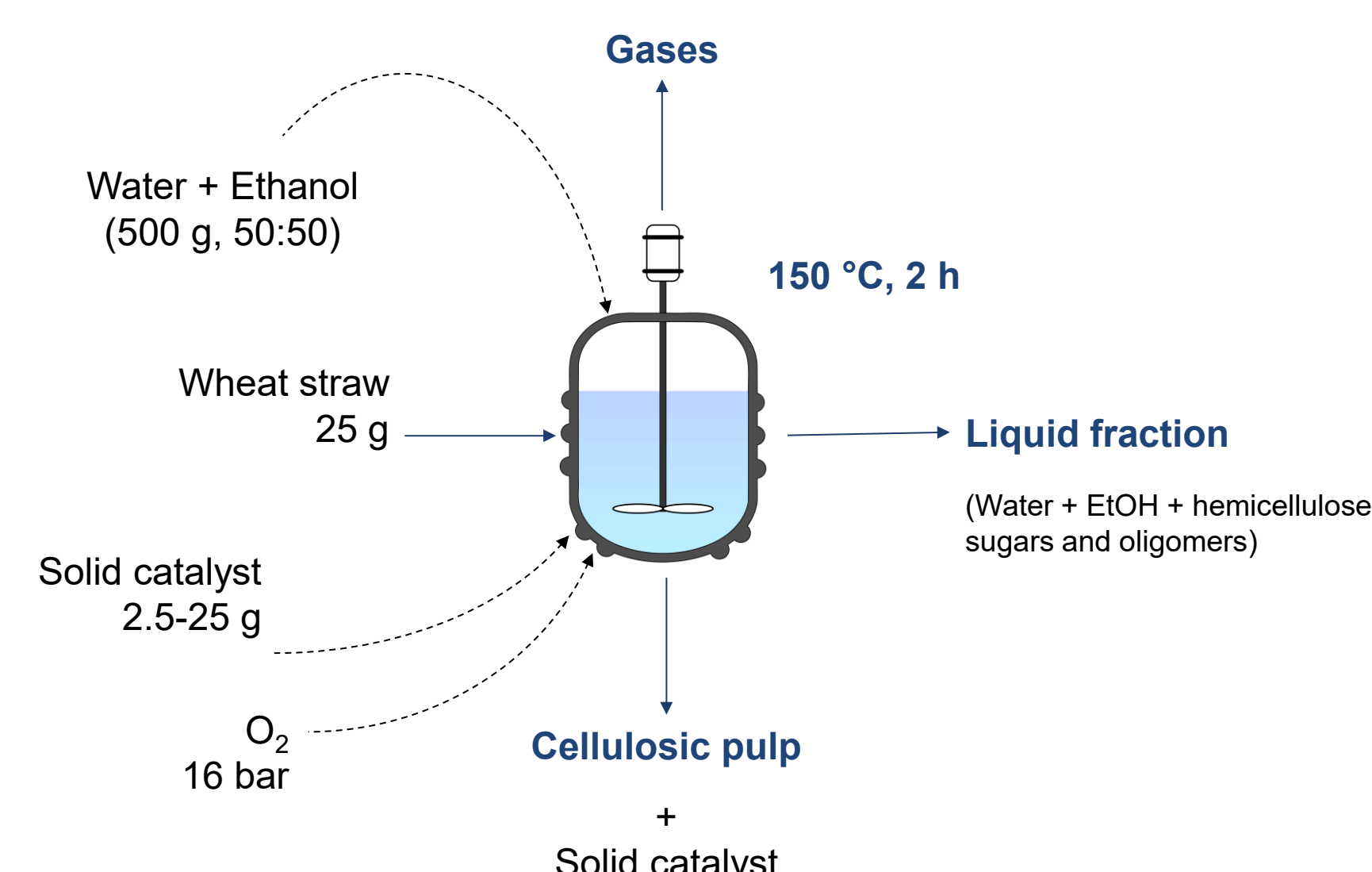


Fig 1. Schematic representation of the OxiOrganosolv experimental setup and procedure.

Product characterisation: NREL method [2] (Pulp composition)

► Acid hydrolysis of pulp followed by ion chromatography (cellulose and hemicellulose content) and UV-Vis spectroscopy (acid-soluble lignin content). Acid insoluble lignin content determined from acid hydrolysis residue.

3 RESULTS

Screening of different zeolites

► The zeolites were screened at 150 °C employing a C/F ratio of 0.1.

► Results were compared to non-catalytic runs at the same temperature.

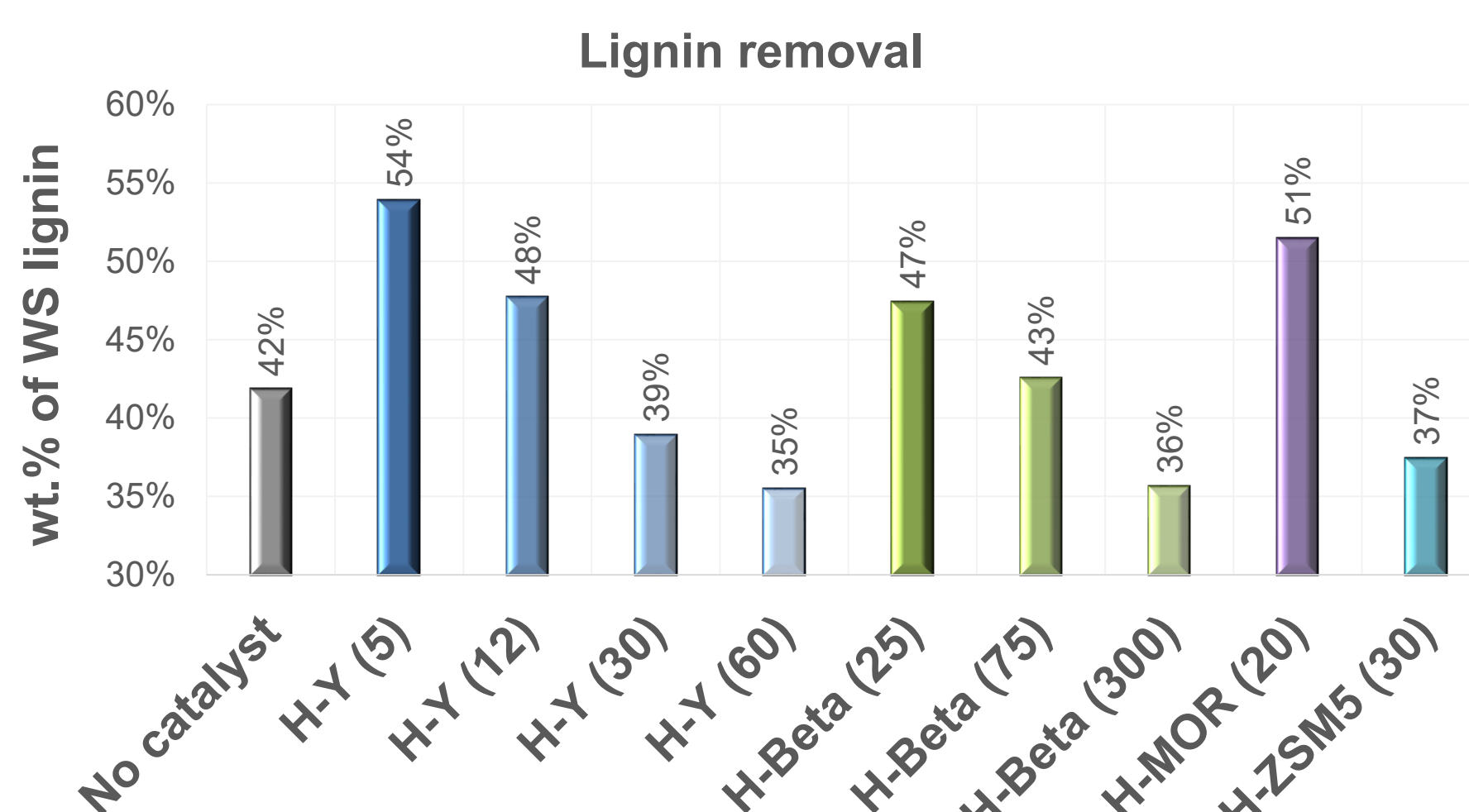
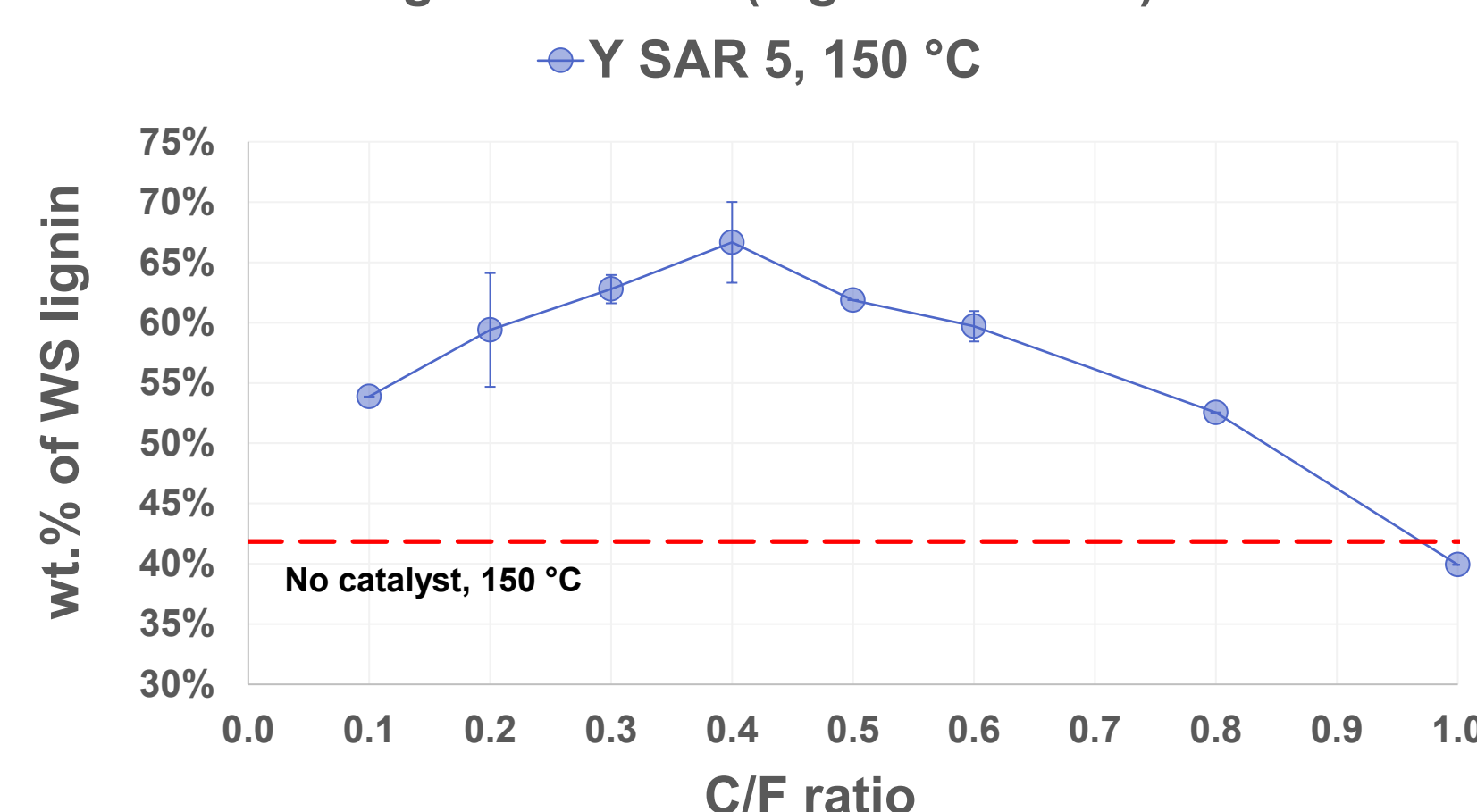


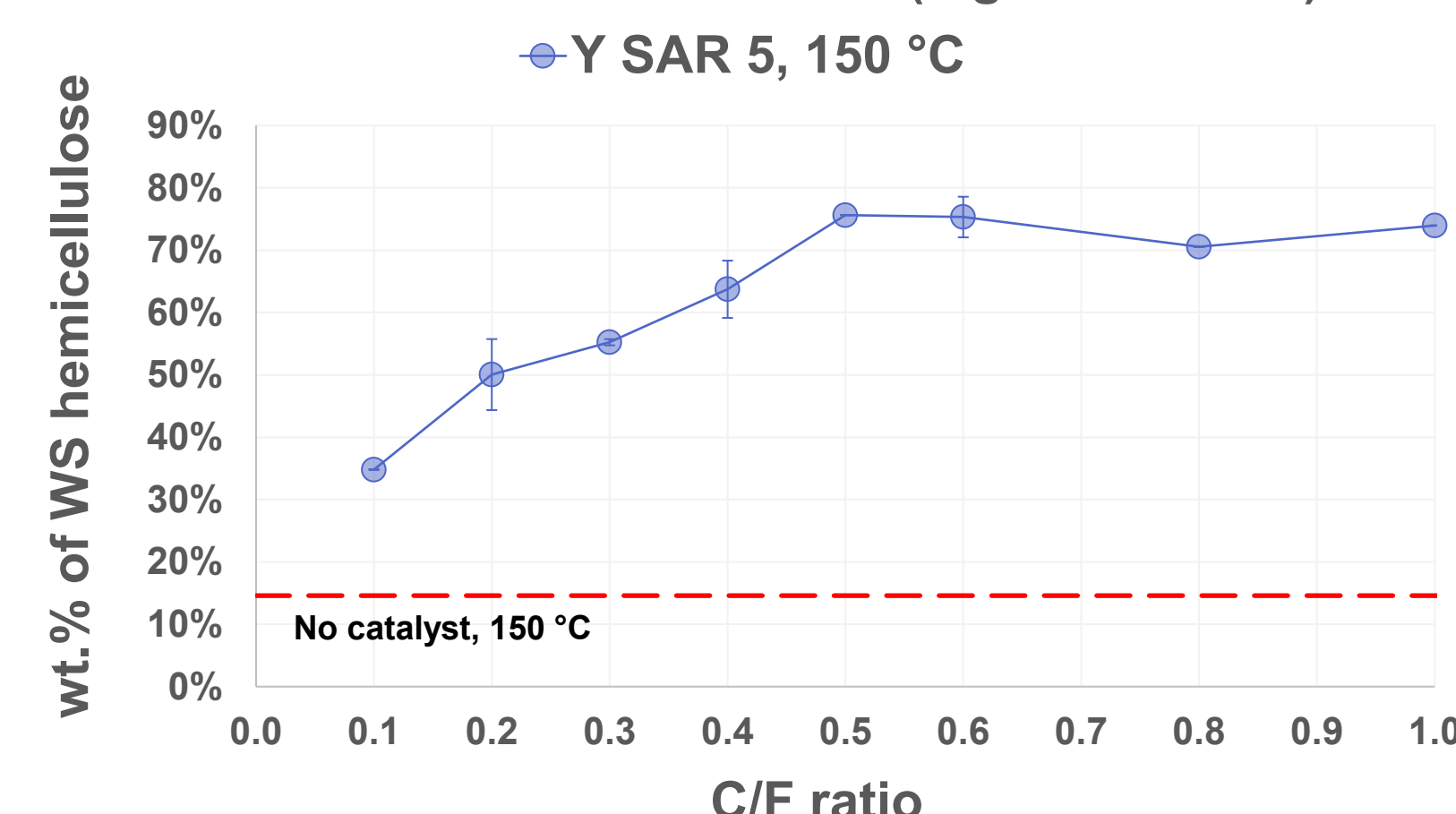
Fig 2. Effect of solid acid zeolites on the degree of lignin removal.

Effect of C/F ratio

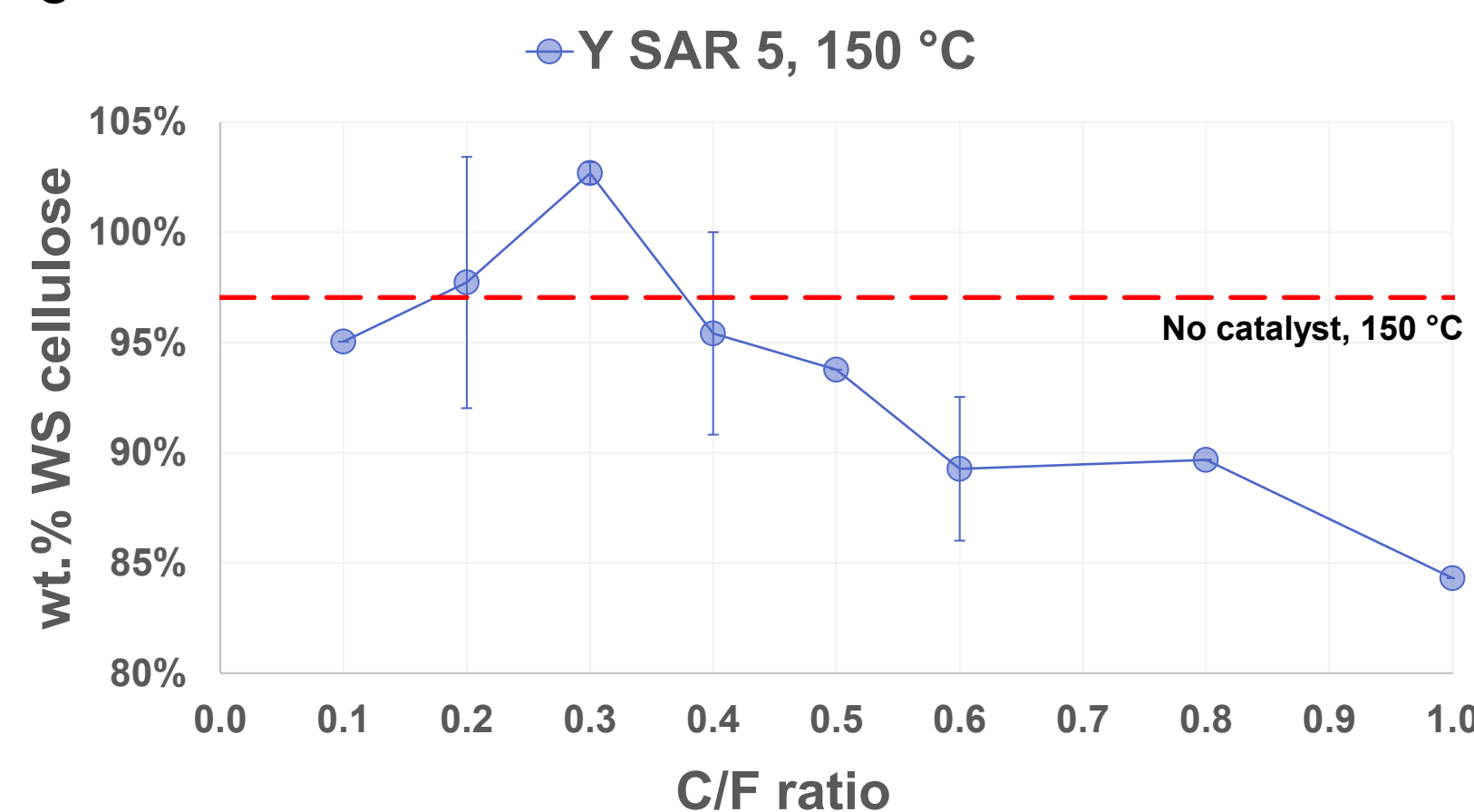
A Lignin removal (higher is better)



B Hemicellulose removal (higher is better)



C Cellulose recovery (higher is better)



D Cellulose in pulp (higher is better)

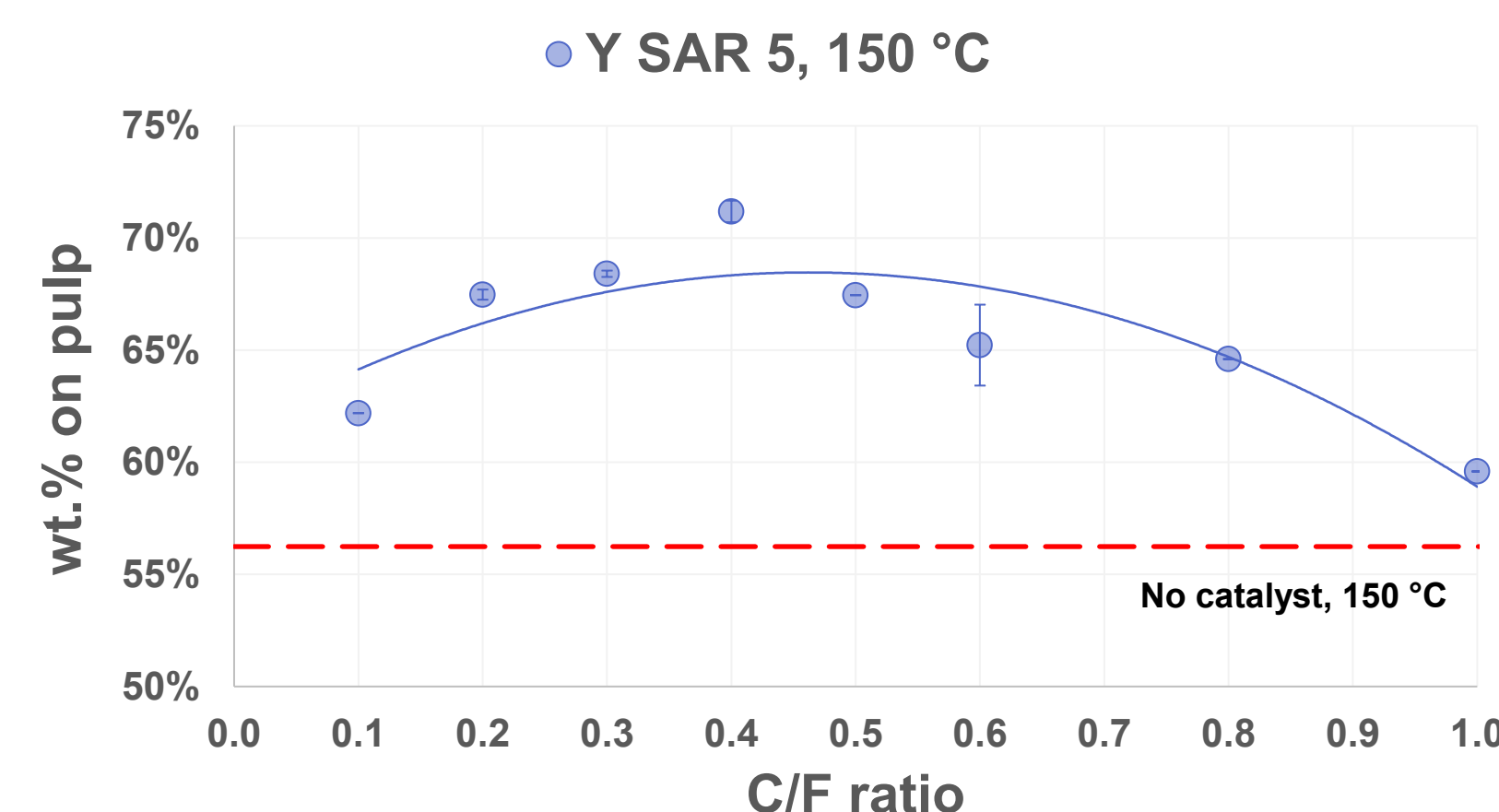


Fig 3. Lignin removal (A), hemicellulose removal (B) and cellulose recovery (C) and cellulose content in the pulp (D) obtained with H-Y(5) as a function of the C/F ratio.

Saccharification potential of pulps

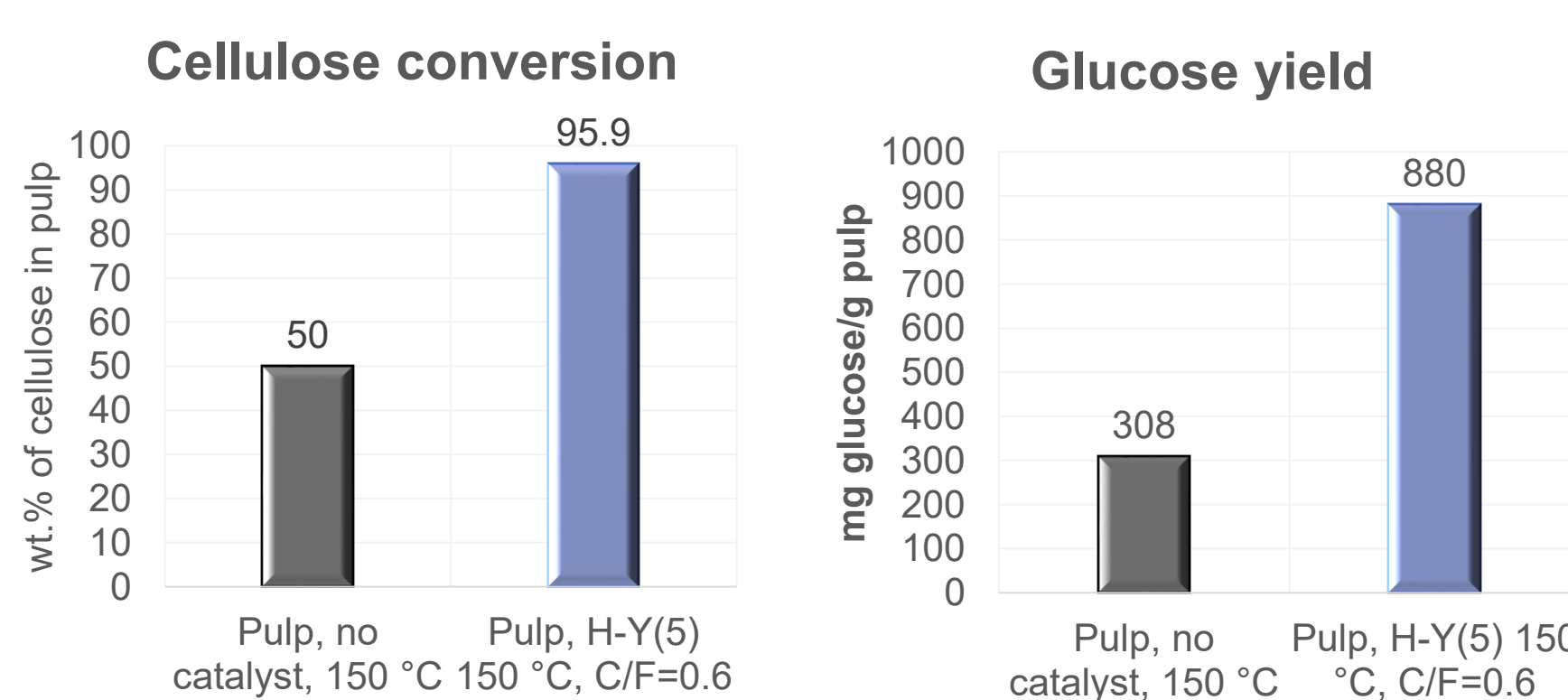
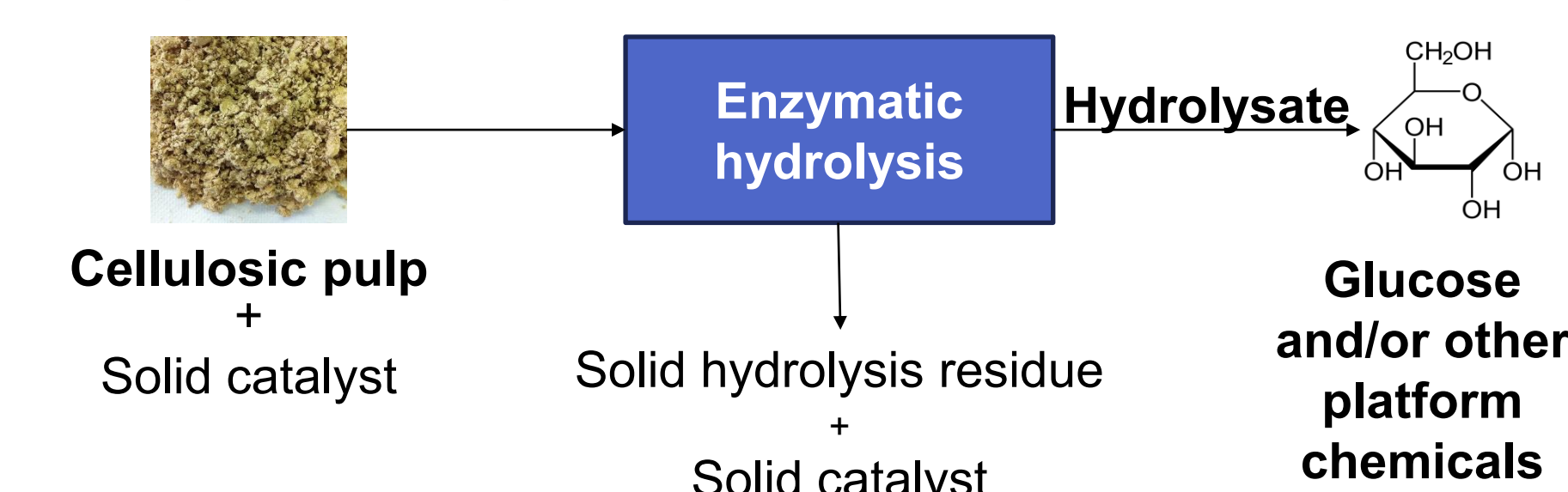


Fig 4. Cellulose conversion (A), and glucose yields (B) obtained from the enzymatic hydrolysis of non-catalytic and catalytic OOS pulps.

Catalyst recovery and characterisation



► The pulp was enzymatically hydrolysed to produce fermentable sugars, and the catalyst was recovered with the hydrolysis residue.

► Organic residues were removed from the catalyst by calcination in air at 600 °C for 3 h.

Table 1. Properties of fresh vs. recovered Y SAR 5 catalysts.

Property	Catalyst	H-Y(5) (fresh)	H-Y(5) (recovered)
Surface area, m ² /g		918	745
Micropore surface area, m ² /g		882	631
Pore volume, cm ³ /g		0.378	0.391
Micropore volume, cm ³ /g		0.333	0.240
Brønsted acid sites, μmol/g		385	211
Lewis acid sites, μmol/g		146	159
K, ppm		0	10,400
Na, ppm		0	12,800
Ca, ppm		0	3,800
Mg, ppm		0	718

4 CONCLUSIONS

► At C/F=0.1, several **zeolites facilitated lignin removal compared to the non-catalytic OOS run** (Fig 2).

✓ Lignin removal correlated with SAR (Fig 2).

► H-Y significantly increased hemicellulose removal due to the **promotion of acid hydrolysis** (Fig 3B).

► Cellulose recovery maintained at about 100% up until C/F=0.4. At higher C/F ratios, cellulose hydrolysis started to occur, reducing recovery (Fig 3C).

► **Pulps with up to 71.2% cellulose content obtained with H-Y(5)** at C/F=0.5 (Fig 3D).

► Enzymatic hydrolysis of the **catalytic OOS pulp** resulted in **higher cellulose conversion and higher glucose yields compared to the pulp obtained from the non-catalytic OOS** (Fig 4).

► The **recovered catalyst** exhibited **reduced surface area and acidity due to its contamination with alkali and alkaline earth metals** (Table 1)

REFERENCES

- [1] K.G. Kalogiannis, et al., **Bioresource Technol.**, 313 (2020), 123599.
- [2] A. Sluiter et al., **Determination of Structural Carbohydrates and Lignin in Biomass**, NREL 2008

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